



# From biomass gasification to large scale liquid biofuels production

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### Lignocellulosic biomass

Agricultural, forest residues wastes



### Pyrolysis (500-800°C)

**Depends:** Method, reactor type, particle size, process parameters  
Low temperature + high residence times → solids  
High temperature + low residence times → liquids

**Yields:** 60-80% w.t. yield bio-oil  
High added value products

**Catalysts:** Zeolites / MCM with Ni, Al, Co, Mo, Fe, Cu

**Challenges:** Composition control, increased H<sub>2</sub>, decreased polymerization rate, thermal control



Catalytic pyrolysis pilot unit, CPERI, Greece

### Activated Carbons

**High added value products**

**Applications:** Gas & liquid cleanup

**Feedstock:** Sugarcane, fruit husks, kernels



Activated carbon

### Gasification (750 - 1200°C)

**Products:** H<sub>2</sub>, CO, CO<sub>2</sub>, CH<sub>4</sub>, C<sub>n</sub>H<sub>m</sub>, Tars, Solids  
Use of air (gas HHV ~ 4-7 MJ/Nm<sup>3</sup>)

**Depends:** Use of H<sub>2</sub>O (gas HHV ~ 15-20 MJ/Nm<sup>3</sup>)  
Temperature, residence time, reactor type, feed technology, use of O<sub>2</sub> or H<sub>2</sub>O

**Gasification:**  $C + O_2 \rightarrow CO_2$      $C + H_2O \rightarrow CO + H_2$   
 $C + CO_2 \rightarrow 2CO$      $CO + H_2O \rightarrow CO_2 + H_2$   
 $C + 2H_2 \rightarrow CH_4$      $2H_2 + O_2 \rightarrow H_2O$

**Pyrolysis:**  $C_mH_n \rightarrow n/4CH_4 + (m+n)/4C$   
 $C_mH_n + (4m-n)/2H_2 \rightarrow mCH_4$

**Reforming:**  $CH_4 + H_2O \rightarrow CO + 3H_2$   
 $CH_4 + CO_2 \rightarrow 2CO + 2H_2$   
 $C_mH_n + mH_2O \rightarrow mCO + (m-n/2)H_2$   
 $C_mH_n + mCO_2 \rightarrow 2mCO + n/2H_2$

### Char oxidation



5 kW<sub>th</sub> pilot scale fluidized bed gasifier, AUTH, Greece

### Syngas Cleaning

**Contents:** particulates, sulphur and alkaline compounds, tars

**Separation:** Metal/ceramic filters (particulates)  
filter selection, material, process conditions  
Sorbents & membranes (S and alkalis)  
sorbent selection, feed technology, fixed bed vessels

**Tars:** Steam reforming/ cracking (tars) ~ catalysts, process conditions  
Tar formation controlled by process conditions, gasifier type and catalyst  
Cold cleaning → Secondary tar removal



Catalytic steam reforming unit, CPERI, Greece

### Bio-oil Reforming

**Separation:** Phenolic fraction → chemicals production  
Hydrocarbonaceous fraction → energy production

**Yield:** Feedstock, composition, H/C fraction, residence time, catalyst  
Lower temperature than methane reforming  
70% conversion to H<sub>2</sub> at 600°C

**Challenges:** Uniform heat supply in catalytic bed  
CO and CO<sub>2</sub> adjusting in the product  
Operating parameters  
Carbon disposition (gasification with H<sub>2</sub>O)

**Technology:** Fluidized bed, catalyst regeneration  
Spray feed at temperatures < 70°C

**Heat needed:** Reforming 125 kg/mol oil  
70-650°C heating (1/4 organics/water) → 319 kJ



Hydrocracking unit  
Source: www.cevron.com

### Fischer-Tropsch Synthesis

**Conditions:** 150-300°C, elevated pressure (promote carbon chain growth)

**Yield:** Mixture of linear paraffins  
Feed composition, reactor design, residence time, catalyst

**Mechanism:**  $(2n+1)H_2 + nCO \rightarrow C_nH_{(2n+2)} + nH_2O$ , n the carbon chain length  
Chain growth follows the ASF model  $W_n/n = (1-a)^2 a^{n-1}$

**Catalyst:** Transition metal based (Fe, Co), optimal H<sub>2</sub>/CO ratio ~ 2

**Challenge:** Need for S free feed (catalyst poisoning)  
Costly equipment, need for constant heat removal



Slurry reactor for Fischer-Tropsch synthesis  
Source: www.sasol.com

### Wax Hydrocracking

**Target:** Narrowing of the F-T paraffin distribution

**Mechanism:** Catalytic isomerization of the long carbon chains and chemical cleavage with H<sub>2</sub>

**Catalyst:** Pt, Pd or bimetallic systems (e.g. Ni/Mo, Ni/W, Co/Mo)

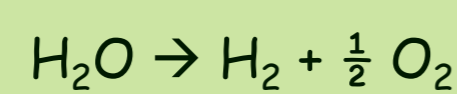
**Technology:** Catalyst and reactor design

### Liquid Biofuels

### Water Electrolysis

Electrolysis using RES (wind and/or solar power)

**Technology:** Electrolysis of water in a PEM



Increased system efficiency and sustainability

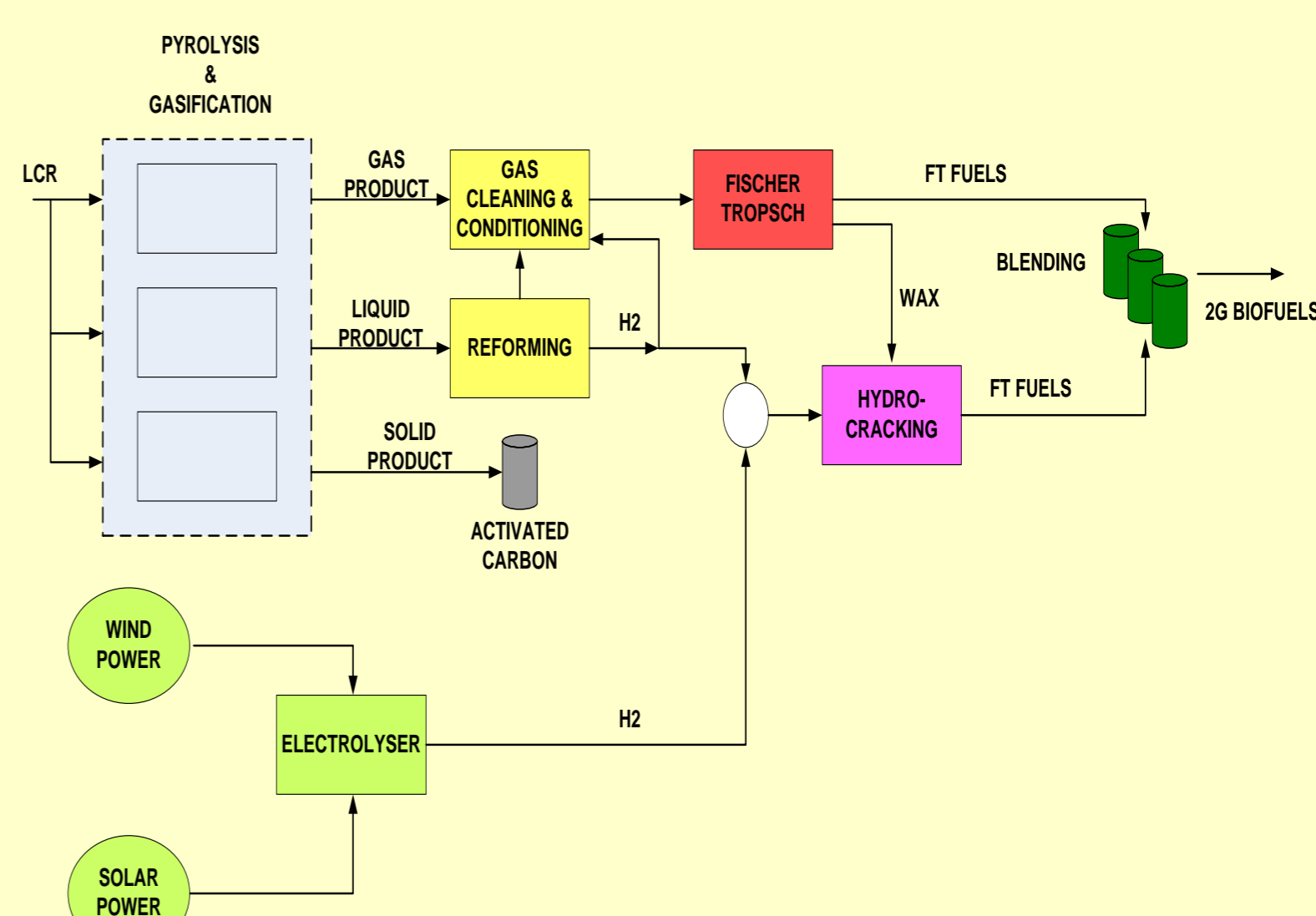
**Challenge:** Detailed energy calculations  
Key meteorological data required



Wind turbines



Solar panels



### Process Integration possibilities - Optimal Design

- Catalysts in the gasifier → tar cracking
- Catalysts in the reformer → max H<sub>2</sub>
- Reformer gas in F-T synthesis (H<sub>2</sub>/CO ~ 2)
- Reformer gas in hydrocracking (H<sub>2</sub>/CO > 2)
- Energy management after syngas cooldown
- Recycle streams in F-T reactor
- Hydrocracking products blending with F-T fuels
- Heat from the F-T reactor

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